Roll No.

Total Pages: 03

BT-2/M-17

8211

CHEMISTRY

CH-101-E

Time: Three Hours]

[Maximum Marks: 100

Note: Attempt *Five* questions in all, selecting at least *one* question from each Unit. All questions carry equal marks.

Unit I

- 1. Discuss the relevance of second law of thermodynamics. (a)
 - (b) Draw the following thermodynamic relationship based on second law:
 - Energy as a function of T and V.
 - Relationship between C_n and C_v . 6,14
- Derive the condition of thermal, mechanical and chemical (a) equilibria for a two-phase system.

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- Discuss the phase diagram of water system. (b)
- Define briefly: (c)
 - Eutectic point
 - (ii) Incongruent m. 1.t.

7,7,6

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P.T.O.

Unit II

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- Write short notes on the following:
 - Reverse osmosis (a)
 - Coagulation (b)
 - Zeolites. (c)
- Discuss the process of sludge formulation along with its 4. (a) composition and properties. How it can be prevented?
 - How many types of hardness of water are there ? (b) Explain, why hard water is unfit for washing and industrial purpose? Explain one method to remove hardness of water.
 - Calculate temporary hardness and total hardness of a (c) sample of water containing $Mg(HCO_3)_2 = 7.3 \text{ mg/L}$; $Ca(HCO_3)_2 = 16.2 \text{ mg/L}, MgCl_2 = 0.5 \text{ mg/L};$ 8,8,4 $CaSO_4 = 13.6 \text{ mg/L}.$

Unit III

- Define Corrosion. Discuss the process of galvanization 5. (a) of iron. How does it prevent the corrosion of iron?
 - State and explain Pilling Bedworth rule. (b)
 - Discuss the mechanism of electrochemical corrosion. (c)
 - Comment on the use of aluminium in place of Zn for (d) 6,6,4,4 cathodic protection of iron from rusting.

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| 6. | Write short notes on the following terms with their significance: | | | | |
|----|--|-------------------------------|----------------|----------------|-------------------------|
| | (a) Water line corrosion | | | | |
| | (b) | Graphite as solid Lubricant | | | |
| | (c) | Flash point | | | |
| | (d) | Lubi | ricant Emulsio | 5,5,5.5 | |
| | | | | Unit IV | |
| 7. | (a) Differentiate between the following: | | | | |
| | | (i) Elastomers and thermosets | | | |
| | | (ii) | Addition an | d condensation | n polymer. |
| | (b) Write the structural monomer unit and structural former of the following: http://www.kuonline.in | | | | and structural formulae |
| | | | | | v.kuonline.in |
| | | (i) | SBR | • | |
| | | (ii) | PVR | | |
| | | (iii) | PF | | • |
| | | (iv) | Silicones. | | 12,8 |
| 8. | Write self explanatory note on the following Analytical methods' techniques: | | | | |
| | (a) Thermogravimetric analysis (b) Conductometric Titrations | | | | |
| | | | | | |
| | (c) Flame Photometry. | | | 7,7,6 | |
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